

Nomenclature Worksheet (a mixture of types)

silver chloride	AgCl	HF	hydrogen fluoride
gold (I) sulfite	Au_2SO_3	FeCl_2	iron (II) chloride
sodium sulfate	Na_2SO_4	$\text{Fe}(\text{NO}_3)_3$	iron (III) nitrate
mercury (I) oxide	Hg_2O	CuSO_4	copper (II) sulfate
potassium phosphate	K_3PO_4	SO_3	sulfur trioxide
manganese (II) chloride	MnCl_2	$\text{Pb}(\text{ClO}_4)_2$	lead (II) perchlorate
lead (II) perchlorate	$\text{Pb}(\text{ClO}_4)_2$	CCl_4	carbon tetrachloride
aluminum hyponitrite	$\text{Al}(\text{NO})_3$	$\text{Ca}(\text{ClO})_2$	calcium hypochlorite
iron (II) phosphate	$\text{Fe}_3(\text{PO}_4)_2$	ZnSO_4	zinc sulfate
copper (II) sulfate	CuSO_4	PbO_2	lead (IV) oxide
magnesium carbonate	MgCO_3	SiO_2	silicon dioxide
zinc bromide	ZnBr_2	CaCl_2	calcium chloride
antimony (III) sulfide	Sb_2S_3	HgI	mercury (I) iodide
carbon monoxide	CO	P_2O_3	diphosphorus trioxide
calcium periodate	$\text{Ca}(\text{IO}_4)_2$	MgSO_3	magnesium sulfite
arsenic (III) iodide	AsI_3	SnCl_4	tin (IV) chloride
tin (II) carbonate	SnCO_3	$(\text{NH}_4)_2\text{SO}_3$	ammonium sulfite
magnesium chlorate	$\text{Mg}(\text{ClO}_3)_2$	$\text{Ba}(\text{OH})_2$	barium hydroxide
silicon dioxide	SiO_2	KBrO_2	potassium bromite
ammonium chloride	NH_4Cl	AgNO_3	silver nitrate
zinc carbonite	ZnCO_2	KIO_2	potassium iodite
gold (I) hypophosphite	Au_3PO_2	HgCl_2	mercury (II) chloride
calcium hydride	CaH_2	$\text{Fe}(\text{NO}_2)_2$	iron (II) nitrite
ammonium sulfite	$(\text{NH}_4)_2\text{SO}_3$	Au_2SO_3	gold (I) sulfite